



## FURFURYL ALCOHOL AS ONE OF THE PRODUCTS OF LIGNOCELLULOSIC



Alexey V. Bykov<sup>1</sup>, Valentina G. Matveeva<sup>1,2</sup> 1 - Tver State Technical University, Tver, 170026, Russia, 2 - Tver State University, Tver, 170100, Russia

matveeva@science.tver.ru

CATALYTIC HYDROPROCESSING

IN OIL REFINING

The increasing interest in biomass as a source of chemicals and energy has recently arisen due to the limited fossil fuel resources and the demand for new clean chemical processes. The research on the development of alternative liquid transport fuels has considerably increased in recent years influenced by the implementation of legislation and directives on alternative energy. To achieve these aims in an ethical, environmental and economical manner the use of lignocellulosic biomass or agri-waste for the production of second-generation biofuels is proposed. Chemicals derived from biomass have recently attracted considerable interest. Furfuryl alcohol is widely used in the chemical industry, mainly for the production of special resins, polymers, and coatings on their basis,

which is resistant to acids, alkalies, and various solvents.

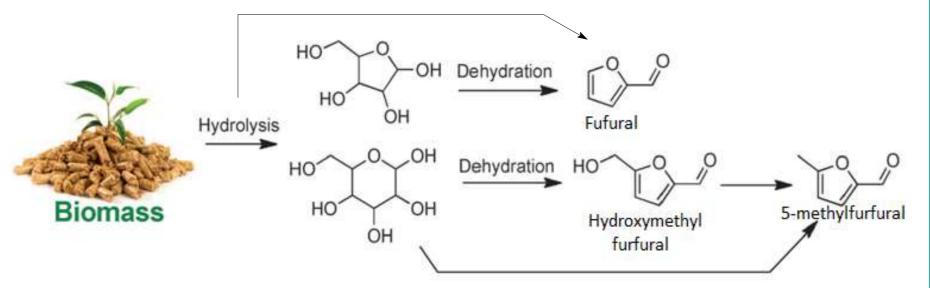


Fig. 1. Acid-catalyzed hydrolysis of biomass and dehydration of sugars to furanic aldehydes

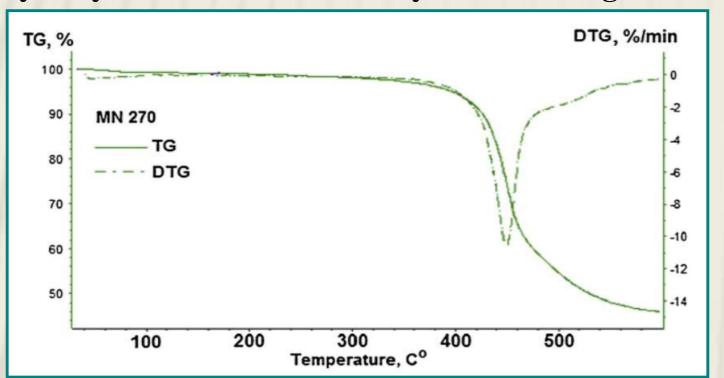
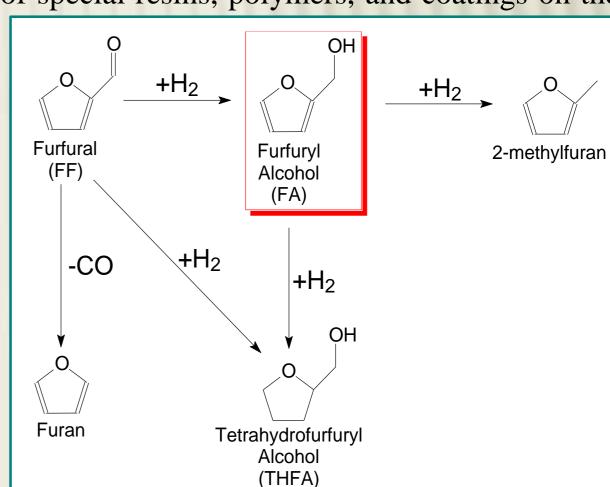


Fig. 2. Results of thermogravimetric analysis of HPS MN 270



Scheme 1. The possible routes of Furfural hydrogenation

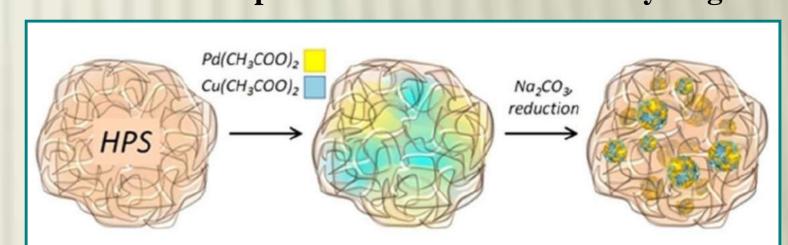
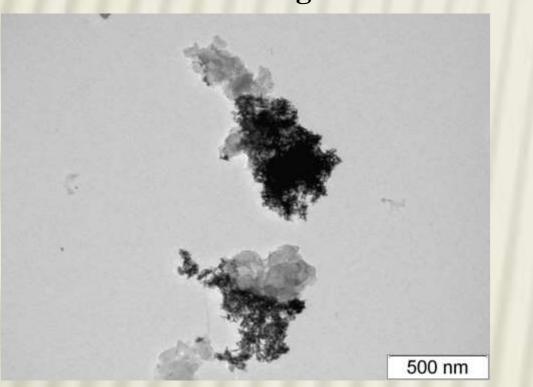


Fig. 6. Schematic representation of the formation of Pd-Cu NPs in the HPS pores



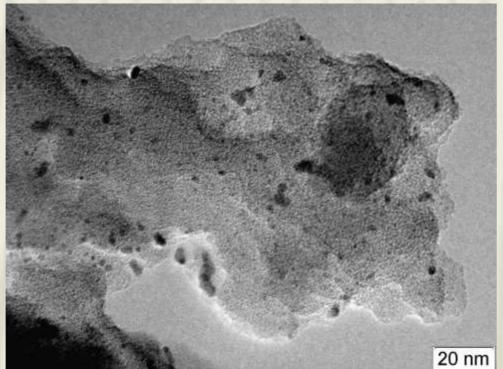


Fig. 3. TEM image of Ru-PS catalyst

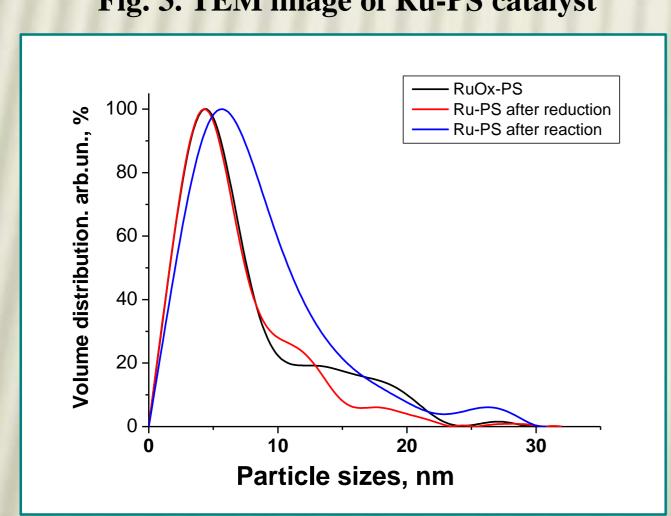


Fig. 4. Volume NP size distribution from the SAXS data

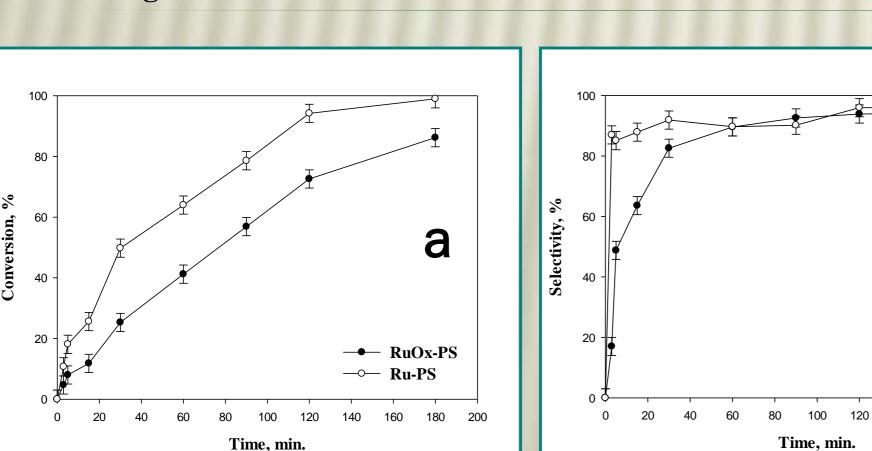
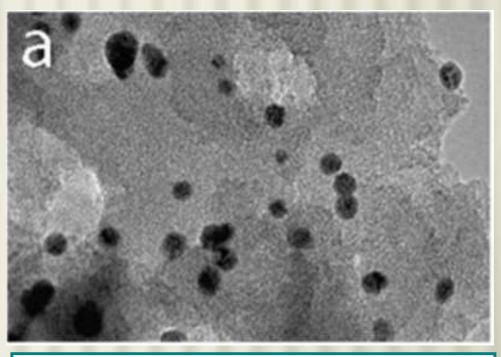
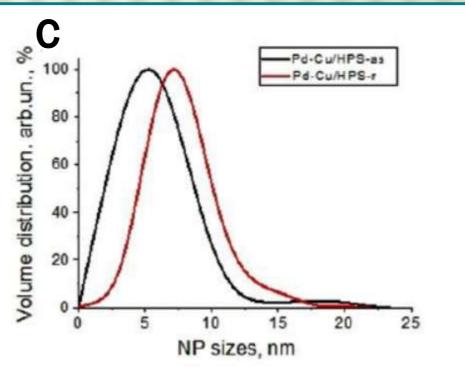
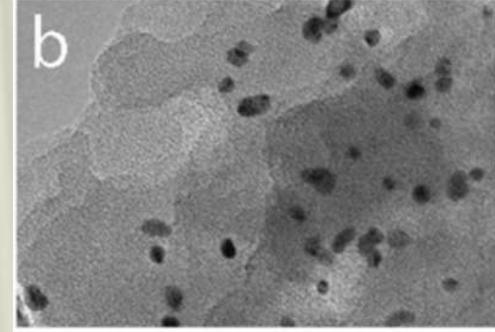


Fig. 5. Dependence of the conversion of furfural (a) and the selectivity of furfuryl alcohol (b) on the reaction time in the presence of ruthenium catalysts







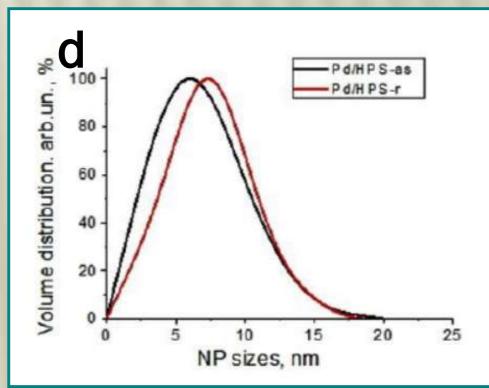
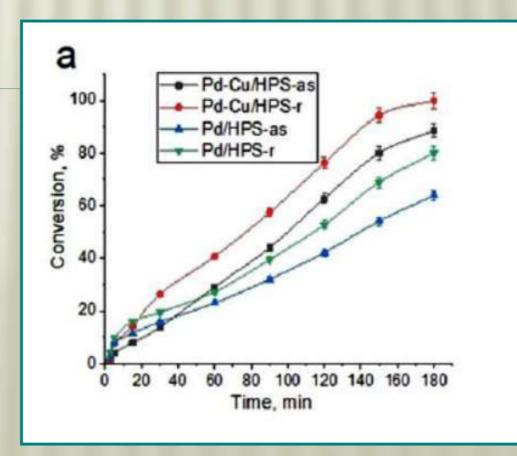


Fig. 7. TEM images (a, b), volume NP size distribution from the SAXS data (c, d)



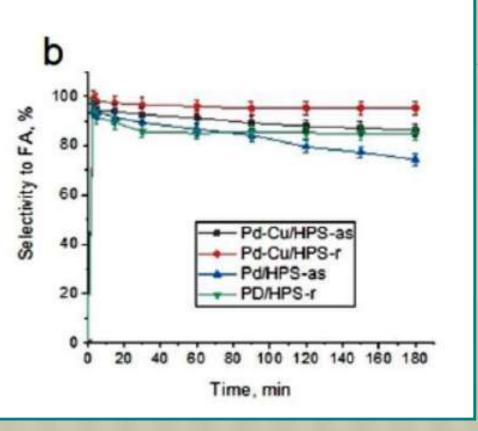


Fig. 8. Dependences of the conversion of furfural (a) and the selectivity of furfuryl alcohol (b) on the reaction time

## Acknowledgments

b

**→** RuOx-PS

The research leading to these results has received funding from the Russian Science Foundation (project 18-19-00240) and from the program "Member of the youth science and innovation

competition" (50098).